



An expeditious synthesis of cyanohydrin trimethylsilyl ethers using tetraethylammonium 2-(carbamoyl)benzoate as a bifunctional organocatalyst

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ARTICLE INFO

Article history:

Received 27 February 2009

Revised 8 April 2009

Accepted 24 April 2009

Available online 3 May 2009

Keywords:

Cyanosilylation

Bifunctional organocatalysis

Tetraethylammonium 2-

(carbamoyl)benzoate

Carbonyl compounds

Cyanohydrins

ABSTRACT

Phthalimide and tetraethylammonium hydroxide react via an unusual pathway to afford tetraethylammonium 2-(carbamoyl)benzoate (TEACB) which is of interest as a bifunctional organocatalyst. TEACB (0.5 mol %) was found to catalyze the addition of trimethylsilyl cyanide (TMSCN) to carbonyl compounds under solvent-free conditions at room temperature with very short reaction times. A wide variety of aldehydes and ketones were transformed into the corresponding cyanohydrin trimethylsilyl ethers in high to quantitative yields.

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Cyanohydrins are highly versatile synthetic intermediates, which can easily be converted into various important building blocks including α -hydroxy acids, α -amino acids, α -hydroxy aldehydes or ketones, β -amino alcohols, and vicinal diols.^{1–5} They are also components of commercially important compounds such as the pyrethroid insecticides, cypermethrin and fluralate.⁵ However, the preparation of cyanohydrins by the addition of highly toxic HCN to carbonyl groups is not a straightforward process and the procedure should be undertaken with caution. The other problem which affects the yields of the products originates in the existence of equilibrium conditions between the reactants and the products.^{3,5a} To overcome these problems, the reaction of trimethylsilyl cyanide (TMSCN)⁶ with carbonyl compounds in the presence of Lewis acid,⁷ Lewis base^{3,8} and double activating^{1c,9} or bifunctional¹⁰ catalytic systems has been described. Since the pioneering works,¹¹ a plethora of catalysts have been reported in the literature for both racemic and asymmetric addition of TMSCN to aldehydes and ketones.^{7–10} The majority of these catalytic systems require metallic Lewis acidic species⁷ which may contain a variety of ligands to enable enantioselective transfer of cyanide to carbonyls.^{1,4a,b,9,10} On the other hand, the recent organocatalytic protocols are particularly attractive because of the mildness of the reaction conditions, operational simplicity, the potential for the development of large

scale or asymmetric versions of the reaction, the ready availability and low toxicity of the organocatalysts.^{3,12–20}

In contrast to the traditional catalytic systems, organocatalytic protocols for cyanosilylation of carbonyl compounds have mainly demonstrated their advantages as Lewis basic catalysts;^{3,12–21} methods using Lewis acidic catalysts are scarce.²² More recently, bifunctional organocatalysts have received attention for cyanosilylation of carbonyl compounds.^{23–25} It is noteworthy that most of the above catalytic systems rely on the activation of only one of the reacting species. Inspired by the multipoint binding sites of enzymes, attention has recently turned to the development of bi- and poly-functional catalytic systems.^{10,22–26} All these developments have advanced significantly the frontier of organocatalytic cyanohydrin synthesis. However, there is still room for improvement, particularly with regard to reaction generality, catalyst simplicity, safer solvents and especially reaction rate. In view of our continuing interest in the use of organocatalysts for efficient cyanosilylation of carbonyl compounds,^{13,15} we herein disclose the first application of tetraethylammonium 2-(carbamoyl)benzoate (TEACB, **1**) as an effective bifunctional organocatalyst for this transformation under solvent-free conditions (Scheme 1).

Potassium phthalimide (PPI) has been traditionally used as a suitable nucleophile in the Gabriel synthesis of primary amines.²⁷ However, it has received less attention as a catalyst in organic synthesis.^{28,29} In our previous works, PPI and the sodium salt of saccharin combined with tetrabutylammonium iodide (TBAI) were shown to be effective Lewis basic organocatalysts for the activation of TMSCN in the cyanosilylation of carbonyl compounds¹⁵ or

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